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Term:

(saline or aqueous buffer) near (iodide salt or
potassium iodide or sodium iodide or iodine salt
or potassium iodine or sodium iodine)

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<u>L16</u>	(saline or aqueous buffer) near (iodide salt or potassium iodide or sodium iodide or iodine salt or potassium iodine or sodium iodine)	1	<u>L16</u>
<u>L15</u>	(saline or aqueous buffer) near (iodide salt or potassium iodide or sodium iodide or iodine salt or potassium iodine or sodium iodine)	1	<u>L15</u>
<u>L14</u>	l10 same l11	4	<u>L14</u>
<u>L13</u>	l12 and ((424/1.11)!.CCLS.)	1	<u>L13</u>
<u>L12</u>	l10 and L11	214	<u>L12</u>
<u>L11</u>	pharmaceutical or pharmaceutically	292288	<u>L11</u>
<u>L10</u>	(saline or aqueous buffer) same (iodide salt or potassium iodide or sodium iodide or iodine salt or potassium iodine or sodium iodine)	331	<u>L10</u>
<u>L9</u>	(saline near aqueous buffer) same (iodide salt or potassium iodide or sodium iodide or iodine salt or potassium iodine or sodium iodine)	0	<u>L9</u>
<u>L8</u>	l6 same (stabilizer or stabilizes or stabilizing)	73	<u>L8</u>
<u>L7</u>	((424/1.11)!.CCLS.) and l6	4	<u>L7</u>
<u>L6</u>	(iodide salt or potassium iodide or sodium iodide or iodine salt or potassium iodine or sodium iodine) same (sodium chloride or potassium chloride or calcium chloride or magnesium chloride)	3926	<u>L6</u>
<u>L5</u>	(polysorbate 80 or tween 80) same (iodide salt or potassium iodide or sodium iodide or iodine salt or potassium iodine or sodium iodine)	23	<u>L5</u>
<u>L4</u>	L1 near (iodide salt or potassium iodide or sodium iodide or iodine salt or potassium iodine or sodium iodine)	5	<u>L4</u>
<u>L3</u>	l2 and (contrast agent or imagining agent)	0	<u>L3</u>
<u>L2</u>	L1 same (iodide salt or potassium iodide or sodium iodide or iodine salt or potassium iodine or sodium iodine)	244	<u>L2</u>
<u>L1</u>	polysorbate 80 or tween 80 or polyethylene glycol or polyethyleneglycol or isooctylphenylether	146850	<u>L1</u>

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Terms	Documents
L7 not 16	2

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<u>L8</u>	L7 not 16	2	<u>L8</u>
<u>L7</u>	L5 and (424/1.49.ccls. or 424/1.61.ccls. or 424/1.65.ccls.)	6	<u>L7</u>
<u>L6</u>	L5 and 424/1.11	8	<u>L6</u>
<u>L5</u>	salt near (stabilizes or stabilizer or stabilizing)	2282	<u>L5</u>
<u>L4</u>	L1 near (stabilize or stabilizer)	42	<u>L4</u>
<u>L3</u>	L1 and (424/1.11 or 424/1.61 or 424/1.65 or 424/1.69)	161	<u>L3</u>
<u>L2</u>	L1 and (imaging or contrast agent)	2366	<u>L2</u>
<u>L1</u>	iodide salt or sodium iodide or potassium iodide	29070	<u>L1</u>

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L8: Entry 61 of 73

File: USPT

Feb 21, 1989

DOCUMENT-IDENTIFIER: US 4806462 A

TITLE: Silver halide photographic material comprising doped divalent metal

Brief Summary Text (18):

The polyvalent metal compound is preferably added in the form of a solution in water or an appropriate solvent, such as methanol, acetone, and the like. In order to stabilize the solution, an aqueous solution of a hydrogen halide, e.g., hydrogen chloride, hydrogen bromide, etc., or an alkali halide, e.g., potassium chloride, sodium chloride, potassium bromide, sodium bromide, etc., can be added. If desired, an acid or an alkali metal may be added. The polyvalent metal compound may be added to a reaction vessel either before grain formation or during grain formation. It may be added to an aqueous solution of a water-soluble silver salt, e.g., silver nitrate, etc., or an alkali halide, e.g., sodium chloride potassium bromide, potassium iodide, etc., and the resulting solution may be added continuously during grain formation. Further, a solution of the polyvalent metal compound may be prepared separately from the water-soluble silver salt or alkali halide and be added continuously at an appropriate stage during grain formation. These methods of addition may be adopted in combinations to advantage.

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